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BYTAUTAS. Chemistry - Section 4 Chemistry

Study of change: Atomic structure of elements, chemical bonding and chemical reactions, molecular structure, electronic configurations and periodic relationships among the elements, thermodynamics, states of matter, and properties of solutions.

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Chemistry Ch 21 Study Guide Electrochemistry Examples, Ch. 21: 11-14, 17-19, 21, 24, 27-30, 59-60, 73 II. Rates of Decay (21.4): Know the half-life equation for radioactive decay (first order reaction) and be able to use it to convert between amount and time, or vice versa. Examples, Ch. 21: 34-42, 80 III. Energy Changes (21.6): Know how to calculate the energy change for a nuclear reaction

~~Chapter 21 - Nuclear Chemistry: Part 1 of 9~~  
~~Chapter 21: Nuclear Chemistry (Chem in 15 minutes or less)~~  
~~Chapter 21 - Nuclear Chemistry: Part 2 of 9~~  
~~Lecture for 12-4-20: Nuclear Chemistry (Chapter 21)~~  
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... Chemistry Honors Ch (1-7.1) -Midterm study Guide Choose the best answer from the options that follow each question. \_\_\_A 1. A scientist is working in the lab, with compounds containing carbon. She is most likely working in the field of The answer is A because organic

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chemistry is the study  
of most carbon-  
containing compounds.

A 2.

this study guide covers  
chapter 13 (chemical  
kinetics), chapter 14  
(chemical equilibrium), and  
chapter 21 (nuclear  
chemistry). all of this  
content will be on the 2nd  
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