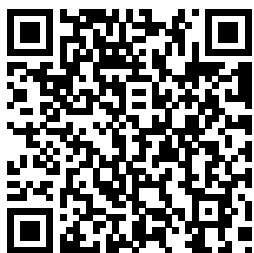


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Major topics:  
reaction rate, rate laws, rate constant,

reaction order, differential vs. integrated rate law, & method of initial rates.

### AP Chemistry

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Equations  
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Answer the  
following  
questions in the  
space provided.  
1. Describe the  
errors made by  
the following  
students in  
making molar  
solutions. a.  
James needs a  
0.600 M  
solution of KCl.  
He measures  
out 0.600 g of  
KCl and adds 1  
L of water to

the solid.  
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Students will have to obtain assistance from the teacher if they need help with the answers.

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Solutions ...

If the rate of

decomposition of

potassium

chlorate at a

certain time is

determined to be

$2.4 \times 10^{-2} \text{ mol}$

$\text{s}^{-1}$ . What is the

rate of formation

of  $\text{O}_2$  at the

same time?

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p7 p8 p9 p10

p11 Unit 4.

Acids & Bases .

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UNIT 4 ACIDS,

BASES AND

SALTS -

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