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the following questions in the space provided. 1. Describe the errors made by the following students in making molar solutions. a. James needs a 0.600 M solution of KCl. He measures out 0.600 g of KCl and adds 1 L of water to the solid.

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If the rate of decomposition of potassium chlorate at a certain time is determined to be $2.4 \times 10^{-2} \text{ mol s}^{-1}$. What is the rate of formation of O_2 at the same time?

Chapter 12 (Chemical Kinetics) - Part 1

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