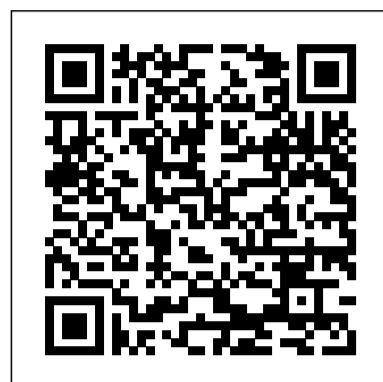


## Chemistry Chapter 12 Stoichiometry

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number of \_\_\_\_ of reactants and products. molecules. All stoichiometric calculations begin with a \_\_\_\_ \_\_\_\_\_. balanced equation. Only \_\_\_\_ and \_\_\_\_ are conserved in every reaction; moles, volumes, and representative particles may not be. mass, atoms. In solving stoichiometric ...

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Chemists use balanced equations to allow them to manipulate chemical reactions in a quantitative manner. Before we look at a chemical reaction, let's consider the equation for the ideal ham sandwich.

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