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explain why changes that lead to greater dispersal of matter are likely to take place. To explain why liquids tend to mix and form solutions. To show how you can predict whether substances are soluble in water.

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Access Free Chemistry Matter And Change Chapter 13 Study Guide Answer Key smallest unit of an element that still maintains the properties that element.

ELEMENT - a pure substance made up of only one type of atom. MATTER and CHANGE Chapter 1 Section 1

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Stoichiometry: Iron is combined with carbon in a series of reactions to form pig iron, which is about 4.3%

carbon. $2C + O_2 \rightarrow 2CO$ $Fe_2O_3 + 3CO \rightarrow 2Fe + 3CO_2$ $2CO + C \rightarrow$ (in iron) CO_2 Pig iron is easier to shape than pure iron, and

the presence of carbon lowers its melting point

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question what is pressure? answer a property of all gas.

It exerts pressure on (pushes against) the sides

question what are the instruments that are used to

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matthewmotamedi. States of Matter. Terms in this set (32)

kinetic energy. The energy an object has because of its

motion. kinetic theory. The tiny particles in all forms of

matter are in constant motion.

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Name Date Class 76 Chemistry: Matter and Change • Chapter 13 Study Guide for Content Mastery Section 13.3 Liquids and Solids In your textbook, read about liquids and solids. In the space at the left, write true if the statement is true; if the statement is false, change the italicized word or phrase to make it true.

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