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STUDY. PLAY. this substance interferes with
hydrogen bonding in water. surfactant.
dissolving medium known as a solution.
solvent. a homogeneous solution of water and
another substance is a called a(n) aqueous
solution. Chemistry Chapter 15 Study Guide
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study guide. Irreversible reaction.

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Match. Gravity. Created by. matthewmotamedi.
Water and Aqueous Systems. Terms in this set
(30) hydrogen bonding. Water's unique and
important properties, including high surface
tension and low vapor pressure, result from
this.

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Chemistry Problems Key Section 15.1 Organic
Compounds 50. Classify each of the
following as organic or inorganic (not
organic) compounds. a. sodium chloride,
NaCl, in table salt inorganic b. hexane,
C₆H₁₄, in gasoline organic c. ethyl
butanoate, CH₃CH₂CH₂CO₂CH₂CH₃, in a
pineapple organic d.

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with hydrogen bonding in water. surfactant.
dissolving medium known as a solution.
solvent. a homogeneous solution of water and
another substance is a called a(n)

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study guide covers chapter 13 (chemical kinetics),
chapter 14 (chemical equilibrium), and chapter 21
(nuclear chemistry). all of this content will be
on the 2nd exam on march 11

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Chemistry Section Goals and Introductions
Section 15.1 Why Solutions Form Goals To
explain why changes that lead to greater
dispersal of matter are likely to take
place. To explain why liquids tend to mix
and form solutions. To show how you can
predict whether substances are soluble in
water.