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16)HZ is a weak acid. An  
aqueous solution of HZ is  
prepared by dissolving 0.020 mol  
of HZ in sufficient water to yield  
1.0 L of solution. The pH of the  
solution was 4.93 at 25.0 ° C.

The  $K_a$  of HZ is \_\_\_\_\_. A)  $9.9 \times 10^{-2}$  ... Chapter 16. Practice  
Questions.tst Author: Satya

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\*According to the Revised  
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this chapter has been  
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### Chemistry Chapter 16 Solutions Practice

Practice-

Problem-6660020 book  
December 22, 2010

8:37 Chapter 16:

Colligative Properties  
of Solutions 45 16-4.

The mole fraction of  
(NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>(aq) is  
given by  $x$  (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>  
 $x = \frac{n(\text{NH}_4)_2\text{SO}_4}{n(\text{NH}_4)_2\text{SO}_4 + n\text{H}_2\text{O}}$

Because (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>(aq) is a strong

electrolyte, it  
dissociates completely  
into NH<sub>4</sub><sup>+</sup> (aq) and  
SO<sub>4</sub><sup>2-</sup> (aq) ions.

Assume a one kilogram  
solution.

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Problems Answers

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Terms in this set (18)

Saturated solution. a solution

containing the maximum amount of solute for a given amount of solvent at a constant temperature and pressure; an equilibrium exists between undissolved solute and ions ...

2020M CHM 112 Chapter 16

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Section 1: Properties of

Solutions 12th solutions of

chemistry in everyday life

class 12 chapter 16 organic

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Chemistry Chapter 16:

Section 2: Concentrations of

Solutions Chemistry Chapter

16 Textbook Questions #19,

21, 34 Video Chapter 16 Acid-

Base Equilibria Chapter 16

Video 1 Chapter 16 Practice

Quiz

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Section 4: Calculations

Involving Colligative

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CHAPTER 16. Colligative Properties of Solutions  
Resonance Formal Charges  
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1; Practice Quiz 2; Practice Quiz 3. Chapter 16 ...  
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Practice Chapter 16 Write if the statement is true and if the statement is false. 1) According to the Arrhenius theory, a neutralization reaction involves the combination of hydrogen ions and hydroxide ions to form water. 1) 2) The term pH 2) 3) The first ionization step is approximately for  $\text{H}_2\text{SO}_3$ .  
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## Chemistry Chapter 16 Chemistry in ...

A "saturated solution" is a solution that "holds the maximum amount of solute" under the given conditions; therefore, no more "solute" can be dissolved into the "solvent" - given that the "temperature and pressure" remain "constant".