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28. CHAPTER 18:

ELECTROCHEMISTRY 665 18.79 The cell voltage is given by: $2 \text{ [Cu] } 0.0257 \text{ V}$ dilute $\ln 2 \text{ [Cu] concentrated} + + = \circ - E$ $E 0.0257 \text{ V } 0.080 0 \ln 2 1.2 = - = 0.035 \dots$

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