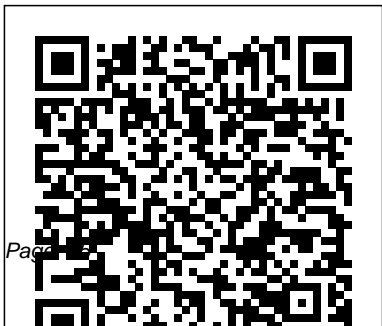


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Chapter 7 Energy and Chemical Reactions - Mark Bishop  
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CHAPTER Section 7.1 continued  
Date Class STUDY GUIDE FOR CONTENT MASTERY STUDY GUIDE FOR CONTENT MASTERY The Elements  
Section 7.1 Properties of s-Block

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Elements In your textbook, read about the representative elements and hydrogen. Answer the following questions. 1. Why are the elements in groups IA—8A called the representative elements?

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Chemistry Chapter 7 Study Guide chemical bond the force that holds two atoms together; may form by the attraction of a positive ion for a negative ion or by sharing electrons

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(Taken from College Board

Sample) Use these answers for

questions 1 - 3. (A) O (B) La

(C) Rb (D) Mg (E) N \_\_\_\_\_ 1.

What is the most

electronegative element of the above? ... AP Chemistry Study Guide – Chapter 7, Atomic Structure and Periodicity

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84 Study Guide for An Introduction to Chemistry Section Goals and

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Energy Goals To introduce the terms energy, kinetic energy, and potential energy.

To introduce the Law of Conservation of Energy. To describe the relationships

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between stability, capacity to do work, and potential energy. To explain why breaking chemical bonds requires energy and why the formation of

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Chapter 7 First Quantum Number principal quantum number, refers to the main energy level, symbol is

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Chemistry - Chapter 7 Study Guide Section 7. 1: Valence electrons electrons in the outer energy level o responsible for bonding with other atoms How do you know how many valence electrons an element has? o Look at the group number Group 1A = 1 valence electron Group 7A = 7 valence electrons Etc

Electron Dot Diagrams a way of showing and keeping track of valence electrons o Written using the ...

