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Section 8.1 Assessment page 247 7. Identify the type of atom that generally forms covalent bonds. The majority of covalent bonds form between nonmetallic elements. 8. Describe how the octet rule applies to covalent bonds. Atoms share valence electrons; the

shared electrons complete the octet of each atom. 9.

Illustrate the formation of single, double, and

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Three indicators of a chemical change include formation of a gas, a change in color/odor, presence of new substances .

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(Part A) Use the periodic table  
to identify each element  
described below. 1.

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