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Characteristics: 1. Generally non-polar (won't dissolve in water) 2. Soluble in non-polar solvents 3. Non-electrolytes. 4. Low melting points ORGANIC CHEMISTRY Continuation of Organic Chemistry I. Methods of analysis, chemistry of [KMBT 654-20150625115832](#)

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Chapter 21: Hydrocarbons

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Chemistry Hydrocarbons Study Guide Answers Compounds Guide Answers Hydrocarbon Compounds Guide Answers Hydrocarbon Compounds Guide Answers A controlled process by which hydrocarbons are broken down or rearranged into smaller, more useful molecules natural gas This is typically made up of 80% methane, 10% ethane, Page 5/29

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Hydrocarbons An infinite variety of compounds can be assembled from only carbon and hydrogen atoms. Such hydrocarbons are the simplest organic compounds, but they are also of prime economic importance because they include the constituents of petroleum and natural gas.

Hydrocarbon Compounds Guide Answers

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diesel fuel that are used in cars, trucks, and buses are hydrocarbons. Organic Compounds Chapter 21: Hydrocarbons Page 8/26 Hydrocarbon Study Guide Answers Hydrocarbon Study Guide Answers Chemistry Hydrocarbons Study Guide Answers Hydrocarbons Study Guide Answers Hydrocarbons are organic molecules that only contain two elements: carbon and hydrogen. Common hydrocarbons include methane, propane, and butane. Section 22hydrocarbon Compound Answer

Chemistry Hydrocarbons Study Guide Answers

Chemistry Study Guide Hydrocarbons Answer

Hydrocarbons Alkanes Study Guide Answers Alkanes have only a single bond in its molecular formula. A hydrocarbon molecule is said to be saturated if there are no double or triple bonds. Alkenes have one or more than one double bond.

Answer true or false: Alkanes are saturated hydrocarbons ...

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the passage. homologous series straight-chain alkanes If all of the carbon atoms are linked by single covalent bonds and there are no branches, the compounds are called (1) examples.

Hydrocarbons Alkanes Study Guide Answers

ChemFacts. • The primary source of hydrocarbons is petroleum. • About 75 million barrels of petroleum are pumped out of the Earth each day. • Hydrocarbons are used as fuels and are the raw materials for products such as plastics, synthetic fibers, solvents, and industrial chemicals.

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