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IDEAL GAS LAW Use the ideal Gas Law below to solve the following problems. pressure in atmospheres volume in liters number of moles L atm Universal Gas Constant = 0.0821

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ric Page 4:2 ANSWER KEY WRITING FORMULAS (CRISS-CROSS METHOD)

Name IONIC BONDING Name \_ IonicbondingoccurswhenametcI

hcnstersonormoreelectrons10anonmelolln on ...

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[NAMING HYDROCARBONS Name](#)

Instructional Fair, Inc. 5. 35 g of  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$  Determine the number of grams in each of the quantities below. 1, 2.5 moles of  $\text{NaCl}$  2. 0.50 moles of 3, 1.70 moles of  $\text{KMnO}_4$  4, 0.25 moles of  $\text{KCl}$  e, ST 5. 3.2 moles of  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$  Chemistry IF8766

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Chemistry IF8766 9 Instructional Fair, Inc. 59. CALCULATIONS USING SIGNIFICANT FIGURES Name ? When multiplying and dividing, limit and round to the least number of significant figures in any of the factors. 11 Example 1 : 23.0 cm x 432 cm x 19 cm = 188,784 cm<sup>3</sup> 11

[SIGNIFICANT FIGURES Name](#)

NAMING HYDROCARBONS Name Name the compounds below according to the IUPAC naming system 1. H H H I I I H -C -C -C -H I I I H H H 5.

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Combined Gas Law Worksheet Answer Key Instructional Fair

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The Mole and Volume Directions: For gases at STP (273K and

1.00 atm pressure), one mole occupies a volume of ...

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Author: ASD Created Date:

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acid + sodium hydroxide Chemistry IF8766 Page 61 Instructional

Fair, Inc. Gas Law Worksheet Answer Details 1. Determining

Molecular Formulas (True Formulas) reagent in analytical chemistry,

dimethylglyoxime, has the empirical This packet is a cumulative review of many topics from the year that are

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1 1 . Sugar dissolves in water. 12. Wood rotting 13. Pancakes cooking on a griddle 14. Grass growing in a lawn 15. A tire is inflated with air. 16. Food is digested in the stomach. 17. Water is absorbed by a paper towel., Inc. Chemistry IF8766 19 @Instructional Fair, Inc . 13.

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[The Mole And Volume Worksheet Instructional Fair Inc](#)

Instructional fair inc chemistry if8766 ideal gas law Solve the

following problems using the ideal gas law: 1) How many moles of

gas does it take. What volume is Indicated on each of the madualed

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Science. a mass pi 10.3 g. If the volume is 58.4 L and the pressure. 758

torrs at a temperature of 2.5° C, 'what\_is the gas? Chemistry 'IF8766 ' 24

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