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Critical Thinking Questions.

4. How many milligrams are there in one kilogram
1,000,000 mg.

5. 21.5 km is how many meters
21,500 m.

6. Is it possible to answer this question: How many mg are in one km Explain. No, because milligrams and kilometers are different units of measure.

7. What is the difference between a Mm and a mm Which is larger ... Use the Guided Inquiry Chemistry

During a chemical reaction, new substances are formed. Reactants

are transformed into different products. Atoms are never created or destroyed, but they are rearranged. A chemical equation represents what happens during a reaction. The following is an example of a chemical equation: Example Equation: $\text{Ca} + \text{HNO}_3 \rightarrow \text{Ca}(\text{NO}_3)_2 + \text{H}_2$

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concept of significant figures has the answer. When multiplying and dividing numbers, you

need to round your answers to the correct number of significant figures. To round correctly, follow these simple steps:

- 1) Count the number of significant figures in each number.
- 2) Round your answer to the least number of

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X = chemical symbol of the element
Z = " atomic number "
A = " mass number " , , and are notations that represent

isotopes of carbon. H, and are notations that represent isotopes of hydrogen. The part of the atom where the protons and neutrons are is called the nucleus. A X Z 12 C 6 C 13 6 C 14 6 12C 6 C 13 6 C 14 6 1H 1 H 2 1 1 H 1 2 H 1 3 3 1

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round your answers to the correct
number of significant figures. To
round correctly, follow these
simple steps: 1) Count the number
of significant figures in each
number. 2) Round your answer to
the least number of significant
figures.

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designed, examine the following

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question 1 and four times your

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the students have not been taught
about the structure of an atom.
They are not familiar with terms
such as atomic number, mass
number or isotopes. Use the

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