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# Chemistry Matter Change Answers Chapter 1

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TEACHER GUIDE AND ANSWERS  
Chemistry: Matter and Change Teacher

Guide and Answers 8 9. The first ionization energies generally increase as you move left-to-right across a period. The increased nuclear charge of each successive element produces an increased hold on the valence electrons. 10. The first ionization energies generally decrease as

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Chemistry Matter And Change Chapter 8 • The partial pressure of a gas depends on the number of moles, size of the container, and temperature and is independent of the type of gas. • At a given temperature and pressure, the partial pressure of 1mol of any gas is the same.

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Type of mixture NOT uniform throughout(clay and water mixture-clay gathers at bottom) Pure substance. Has fixed composition and differs from a mixture in that: 1- every sample has exact same characteristic physical and chemical properties used to identify substance. 2- every sample of a pure substance has exact same composition with the same makeup (i.e.: pure water is always 11.2% H and 88.8% oxygen by mass.  
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the statement is false, change the italicized word or phrase to make it true 8 To determine the  
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1 Chapter 2.1, 2.2 Review Packet – Answer Key 2.1 The Nature of Matter Lesson Summary Atoms The atom is the basic unit of matter, made up of three subatomic particles. Protons have a positive charge and neutrons carry no charge. Strong forces bind protons and neutrons together in the nucleus.

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