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Calculate the heat required to melt 25.7 g of solid ...

Figure 15.4.1 Energy Changes Accompanying the Thermite Reaction

Because enthalpy is a state function, the overall enthalpy change for the reaction of 2 mol of Al(s) with 1 mol of Fe₂O₃(s) is -851.1 kJ, whether the reaction occurs in a single step (H₄, shown on the left) or in three hypothetical steps (shown on the right) that involve

...

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