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Chemistry: Matter and Change Chapter 5 c. d. Name CHAPTER Applying Scientific Methods Date Class CHAPTER ASSESSMENT A chemist isolated four samples, A, B, C, and D. She obtained the following atomic emission spectra of the samples. 400 500 600 700 29 Nanometers
1. Examine each sample's atomic emission spectra.
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Chemistry: Matter and Change Chapter 4 . Name CHAPTER Section 4.3 continued Date Class STUDY GUIDE FOR CONTENT MASTER 16. An isotope has an atomic number 51 and a mass number 123. Answer the following question. 17. Which of the isotopes in problems 13–16 are isotopes of the same element?

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Videos; Chapters 1 and 3 of Matter pages 70-75 Hetero) 3.4 Pure
 3. Chapters 1 & 3 Study Problem-Solving Lab 1. Substances (Elements
 Guide; Chapters 1 & 3 Explain why the flow of & Compounds)
 Outline; Chapter 2 a compressed gas must
 Analyzing Data. Chapter be controlled for
 Assessment - Chapter 2; practical and safe use.
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 Chemistry Conversion change chapter 3
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 Section 3.1 Properties

Chemistry Matter and
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 (Matter - Properties
 & Changes) 3.1
 Properties of Matter
 (Physical & Chemical)
 3.2 Changes in Matter
 (Physical & Chemical)
 3.3 Mixtures of
 Matter (Homo &