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Section 9.1 Reactions and Equations In your textbook, read about evidence of chemical reactions. For each statement, write yes if evidence of a chemical reaction is present. Write no if there is no evidence of a chemical reaction. 2. 4. 5. 6. 8. A tomato smells rotten. A drinking glass breaks into smaller pieces.

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STUDY GUIDE Section 7.3
continued For each of the following chemical formulas, write the correct name of the ionic compound represented. You may refer to the periodic

table on pages 156–157 and Table 8.7 for help. 19. ...
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Name Date Class Study Guide for Content Mastery Chemistry: Matter and Change • Chapter 3 13

Matter—Properties and Changes Matter—Properties and Changes. Section 3.1 Properties of Matter. In your textbook, read about physical properties and chemical properties of matter. Use each of the terms below just once to complete the passage.

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Section 4.4 Changes to the Nucleus—Nuclear Reactions In your textbook, read about radioactivity. For each item in Column A, write the letter of the matching item in Column B. Column A 1. The rays and particles that are emitted by a radioactive material 2. A reaction that involves a change in an atom's nucleus 3.

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STUDY GUIDE FOR CONTENT

MASTERY The Periodic Table and Periodic Law Section 6.1 Development of the Modern Periodic Table In your textbook, reads about the history of the periodic table's development. Use each of the terms below just once to complete the passage. nine eight accepted octaves elements protons atomic mass properties periodic law [Chemistry Change And Matter Study Guide Answers](#)

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Matter—Properties and Changes Matter and Change Chemistry Study Guide. A series of steps followed to solve problems including collecting data, formulating a hypothesis, testing the hypothesis, and stating conclusions.

Chapter 4 Chemistry: Matter and Change (Study Guide ...

TEACHER GUIDE AND ANSWERS

Chemistry: Matter and Change

Teacher Guide and Answers 8 9.

The first ionization energies generally increase as you move left-to-right across a period. The increased nuclear charge of each successive element produces an increased hold on the valence electrons. 10. The first ionization energies generally decrease as

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Bonding Section 9.2 Naming

Molecules In your textbook, read about how binary compounds and acids are named from their formulas.

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The Mixtures and Solutions chapter of this Glencoe Chemistry - Matter and Change companion course helps students learn the essential chemistry lessons of mixtures, solvents and solutes.

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Chapter 4 Chemistry: Matter and
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extremely small, positively charged,
dense center of an atom that
contains positively charged protons,
neutral neutrons, and is surrounded
by empty space through which one
or more negatively charged
electrons move.

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