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periodic table are arranged according to increasing (8) , as a result of the work of (9) . This arrangement is based on number of (10) in the nucleus of an atom of the element. The modern form of the periodic table results in the (11) which states that when elements are arranged according to increasing atomic number, there is' a periodic repetition of their chemical and physical (12) Study Guide for Content Mastery Chemistry: Matter and Change Chapter 6. Matter and

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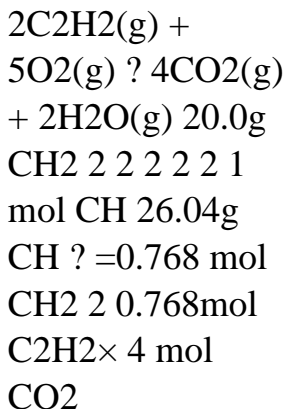
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Compounds, Classification of Matter, Chemistry Examples, Know This For Your Chemistry Final Exam - Stoichiometry Review Matter Compilation: Crash Course Kids Chemistry Test Ch 1. Study Guide. Matter and Change. Define Matter. States of matter. Phase changes (including graphs) Identify the characteristics of a substance. Compare and contrast mixtures and substances. Classify mixtures as homogeneous or heterogeneous.
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Bonding Section 9.2

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about how binary
compounds and acids
are named from their
formulas.

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and Answers 7

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Chapter 12 –

States of Matter

Section 12.1

Gases 1. motion 2.

a. small b. forces

c. random d.

elastic; kinetic 3.

KE $1/2 mv^2$ 4.

Temperature 5.

true 6. true 7. false

8. true 9. true 10.

false

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and Change

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Matter Notes

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Notes IGCSE #2

IGCSE CH 2

Mixtures-

Separation Methods

– Vocabulary

Matter Notes #3

Pure Substances

and Mixtures

Matter Notes #4 –

Separation Methods

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IGCSE A volatile

History assignment

Matter and Change

Chemistry Study

Guide. STUDY.

Flashcards. Learn.

Write. Spell. Test.

PLAY. Match.

Gravity. Created by.
NATHAN_SINK.

Terms in this set
(48) Chemistry. the
study of matter and
how it changes.

Matter. Anything
that has mass and
takes up space.

Mass. A measure of
the amount of
matter in an object.