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ANSWERS. Structures F and G are achiral.

The former has a plane of symmetry passing through the chlorine atom and bisecting the opposite carbon-carbon bond. The similar structure of compound E does not have such a symmetry plane, and the carbon bonded to the chlorine is a chiral center (the two ring segments connecting this carbon are not identical).). Structure G is essentiall

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Problems Answer 11-2 Practice Problems 1

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Titration of a strong acid with a strong base

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