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Note : # Neutrons = Mass
Number – Atomic Number. #
Neutrons of Nitrogen-14 = 14
- 7 = 7. # Neutrons of
Nitrogen-13 = 13 - 7 = 6. c)

The number of electrons in a
neutral atom always equals the
atomic number. The atomic
number of both isotopes is 7,
therefore the number of
electrons in both isotopes is
also 7.

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below: According to question:
Mass of ethylene glycol = 120
g Mass of liquid = 450 g
Therefore, percentage...
During the doctor visit of
Person G; he complained
about feeling tired.

(Northern Arizona University)
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