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[Experiment and Theory](#) Wiley

The eleventh edition was carefully reviewed with an eye toward strengthening the content available in OWLv2, end-of-chapter questions, and updating the presentation. Nomenclature changes and the adoption of IUPAC periodic table conventions are highlights of the narrative revisions, along with changes to the discussion of d orbitals. In-text examples have been reformatted to facilitate learning, and the accompanying Interactive Examples in OWLv2 have been redesigned to better parallel the problem-solving approach in the narrative. New Capstone Problems have been added to a number of chapters. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version.

[High School Chemistry with Regents Exams - The Physical Setting](#) Cengage Learning

The Study Guide reflects the unique problem-solving approach taken by the Chemical Principles text. The new edition of the Study Guide includes many new worked out examples.

Elsevier

This Second Edition of the first-year chemistry text known for its clarity of exposition and its large number of illustrative worked problems, contains a more rigorous treatment of electrochemistry, chemical equilibrium, and thermochemistry. Worked examples now number over 300, and exercises, over 1460.

["O" Level Study Guide - Chemistry Quite Easily Done](#) Hodder Education

This survival guide focuses on helping students practice for exams and shows them how to solve difficult problems by dissecting them into manageable chunks. Written in the style of a student meeting with an instructor during office hours, it addresses the most frequently asked questions. This approach leads to the three levels approach - A, B, and minimal - to dissect a typical difficult question into manageable chunks and quickly build student confidence to master the knowledge needed to succeed in the course. This book is available for students to purchase at www.CENGAGEbrain.com or available for packaging with any Cengage textbook. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version.

[The Practice of Chemistry Study Guide & Solutions Manual](#) Macmillan

This is the revised edition of the best-selling chemistry textbook for majors in allied health fields, emphasizing the molecular basis of life. Sound treatment of fundamentals is supported by examples from DNA and genetic engineering, radioimmunology, the selection and use of radioisotopes in medicine, biometallic corrosion of metal alloys, medical emergencies of acid-base blood chemistry, and neurotransmitters and drugs of the central nervous system. Features new chapters on biochemistry and a consolidated discussion of stoichiometry. Technical terms are carefully defined and consistently used. Exercises and marginal comments further clarify concepts.

[Survival Guide for General Chemistry with Math Review and Proficiency Questions: How to Get an A](#) Cengage Learning

Written for general chemistry courses, 'Chemical Principles' helps students develop chemical insight by showing the connection between chemical principles and their applications.

[Study Guide to Accompany Calculus for the Management, Life, and Social Sciences](#) Houghton Mifflin

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[Chemical Principles Student's Study Guide & Solutions Manual](#) Wiley

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[Know Your 'O' Level Chemistry - A Study Guide](#) John Wiley & Sons

Study Guide to Accompany Basics for Chemistry is an 18-chapter text designed to be used with Basics for Chemistry textbook. Each chapter contains Overview, Topical Outline, Skills, and Common Mistakes, which are all keyed to the textbook for easy cross reference. The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms, a summary of general concepts, and a list of numerical exercises, while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook. The Fill-in, Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples. The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry, while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter. After briefly dealing with an overview of chemistry, this book goes on exploring the concept of matter, energy, measurement, problem solving, atom, periodic table, and chemical bonding. These topics are followed by discussions on writing names and formulas of compounds; chemical formulas and the mole; chemical reactions; calculations based on equations; gases; and the properties of a liquid. The remaining chapters examine the solutions; acids; bases; salts; oxidation-reduction reactions; electrochemistry; chemical kinetics and equilibrium; and nuclear, organic, and biological chemistry. This study guide will be of great value to chemistry teachers and students.

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[A Level Chemistry Multiple Choice Questions and Answers \(MCQs\)](#) American Water Works Association

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The book itself contains chapter-length subject reviews on every subject tested on the AP Chemistry exam, as well as both sample multiple-choice and free-response questions at each chapter's end. Two full-length practice tests with detailed answer explanations are included in the book.

Study Guide to Accompany Basics for Chemistry, David A. Ucko John Wiley & Sons

This is an ebook version of the "Advanced Study Guide - Chemistry - Ed 1.0" published by Step-by-Step International Pte Ltd. [For the Higher 2 (H2) syllabus with last exam in 2016.] This ebook gives concise illustrated notes and worked examples. It is organised largely accordingly to the Singapore-Cambridge GCE A-Level Higher 2 (H2) syllabus, with additional topics to cover the equivalent syllabuses of the University of Cambridge International Examination (CIE) A Level (Core & A2), and the International Baccalaureate (IB) Higher Level (Core & AHL). The concise notes cover essential steps to understand the relevant theories. The illustrations and worked examples show essential workings to apply those theories. We believe the notes and illustrations will help readers learn to "learn" and apply the relevant knowledge. The ebook should help readers study and prepare for their exams. Relevant feedbacks from Examiner Reports, reflecting what the examiners expected, are incorporated into the notes and illustrations where possible, or appended as notes (NB) where appropriate. It is also a suitable aid for teaching and revision. Sample pages are available (in .pdf) from our website.

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Study Guide for Whitten/Davis/Peck/Stanley's Chemistry, 10th Step-by-Step International Pte. Ltd.

The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon, as it forms around particles of a metal catalyst. Carbon nanotubes are a recently discovered allotrope of carbon. Three other allotropes of carbon-buckyballs, graphite, and diamond-are illustrated at the left, as is the molecule methane, CH₄, from which nanotubes and buckyballs can be made. The element carbon forms an amazing number of compounds with structures that follow from simple methane, found in natural gas, to the complex macromolecules that serve as the basis of life on our planet. The study of chemistry also follows from the simple to the more complex, and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement.