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listing of terms, a summary of general concepts, and a list of numerical exercises, while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook. The Fill-in, Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples. The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry, while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter. After briefly dealing with an overview of chemistry, this book goes on exploring the concept of matter, energy, measurement, problem solving, atom, periodic table, and chemical bonding. These topics are followed by discussions on writing names and formulas of compounds; chemical formulas and the mole; chemical reactions; calculations based on equations; gases; and the properties of a liquid. The remaining chapters examine the solutions; acids; bases; salts; oxidation-reduction reactions; electrochemistry; chemical kinetics and equilibrium; and nuclear, organic, and biological chemistry. This study guide will be of great value to chemistry teachers and students.

Barron's Science 360: A Complete Study Guide to Chemistry with Online Practice Hodder Education

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Fundamentals of General, Organic, and Biological Chemistry, Study Guide and Selected Answer Manual John Wiley & Sons

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The SOLARO Study Guide is designed to help students achieve success in school. It is a complete guide to be used by students throughout the school year for reviewing and understanding course content, and for preparing for assessments. The content in the California High School Chemistry study guide is 100 percent curriculum aligned and serves as an excellent source of material for review and practice. Each Class Focus includes the following sections: the Periodic Table; Chemical Bonds; Conservation of Matter and Stoichiometry; Gases; Acids and Bases; Solutions; Chemical Thermodynamics; Reaction Rates; Chemical Equilibrium; Organic Chemistry; and Nuclear Processes. To create this book, teachers, curriculum specialists, and assessment experts have worked closely to develop the instructional pieces that explain each of the key concepts for the course. The practice questions and sample tests have detailed solutions that show problem-solving methods, highlight concepts that are likely to be tested, and point out potential sources of errors. Enhanced treatment of concepts, more practice sections, and additional learning tools are found in the accompanying digital version of SOLARO which may be accessed through the web or on mobile devices.

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Fundamentals of Chemistry, Study Guide John Wiley & Sons

The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon, as it forms around particles of a metal catalyst. Carbon nanotubes are a recently discovered allotrope of carbon. Three other allotropes of carbon—buckyballs, graphite, and diamond—are illustrated at the left, as is the molecule methane, CH₄, from which nanotubes and buckyballs can be made. The element carbon forms an amazing number of compounds with structures that follow from simple methane, found in natural gas, to the complex macromolecules that serve as the basis of life on our planet. The study of chemistry also follows from the simple to the more complex, and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of

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