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Hint: add your answer to question 1 and four times your answer to question 2. $4 + 4(7) = 4 + 28 = 32$ 4. Verify that there are 32 electrons pictured in the drawing of CCl_4 . Count each dot in the picture above: there are 24. There are also 8 more electrons represented by the 4 lines (each line represents two electrons). 5.

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Unit 8: More Chemical Reactions

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SslgDcll?g Ealue - LTHS Answers

Instructions on balancing chemical equations: Enter an equation of a chemical reaction and click 'Balance'. The answer will appear below; Always use the upper case for the first character in the element name and the lower case for the second character.

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Introduction Questions Information: Subscripts and ...

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Balancing Equations: Practice Problems

ChemQuest 29 Name: _____ ... Introduction Questions 1. Write the equation for the reaction of sodium and chlorine (diatomic) to form sodium chloride. 2. Write the equation for the reaction of calcium nitride and sodium chloride to produce calcium chloride and sodium nitride. ... balancing an equation containing polyatomic ions is really not ...

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ChemQuest 37 SslgDcll?g E"alue Date Hour I ntroduction Questions Information: Subscripts and Coefficients A subscript is a small number that tells you how many atoms are in a compound.For

example, in $CaCl_2$ the two is the subscript and it tells us that there are two chloride ions bonded to one calcium. A coefficient tells also tells us how many atoms or compounds there are, but in a different way.

ChemQuest 23 : Drawing Covalent Compounds

86 ChemQuest 29 Name: _____ Date: _____ Hour: _____ Introduction Questions 1. Write the equation for the reaction of sodium and chlorine to form sodium chloride. $Na + Cl_2 \rightarrow NaCl$ 2. Write the equation for the reaction of calcium nitride and sodium chloride to produce calcium chloride and sodium nitride.

63 ChemQuest 18 Electron Configurations66 ChemQuest 19 ...

Balancing Equations: Answers to Practice Problems 1. Balanced equations. (Coefficients equal to one (1) do not need to be shown in your answers).