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### Concentration Answers

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1) If I make a solution by adding 83 grams of sodium hydroxide to 750 mL of water...

To solve problem 1, you need to have calculated for various parts that there are 2.08 moles of NaOH (which has a molar mass of 40 g/mol), that there are 750 grams of

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Answers Show all work and circle the final answer. 1. Determine the pH of a 0.010 M HNO

3 ... Concentration Review Worksheet -

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solute/kg of solvent). 5] 8.2 moles of ephedrine in 650 mL of isopropyl alcohol (density = 0.7855 g/mL) (650 mL)(0.7855 g/mL) = 510.6 g => 511 g.

### Concentration Review Worksheet Answers

Concentration Review Worksheet. 1) If I make a solution by adding 83 grams of sodium hydroxide to enough water to make 750 mL of solution... a) What is the molarity of this sodium hydroxide solution? b) What is the percent by mass of sodium hydroxide in this solution? c) What is the density of sodium hydroxide

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in this solution?

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