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11.1 Describing Chemical Reactions 11 - Evaluation 2016

Valence and core Electrons Name Date O O Purpose To discover the arrangements of electrons within atoms. Complete the table on the handout, filling in the missing atoms. Then answer the questions. 1. How does the number of electrons change as you move from left to right a period? ...

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Choose the best answer and write its letter on the line. 1. Which of the following tasks would probably not be assigned to an ... Discuss the role of chemistry in one of these areas: Materials, Energy, Medicine and Biotechnology, Agriculture, the Environment, or the Universe. ... Core Teaching Resources .

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Core Teaching Resources . Name Date Class IONIC AND METALLIC BONDING Practice Problems In your notebook, answer the following SECTION 7.1 IONS For each element below, state (i) the number of valence electrons in the atom, (ii) the electron dot structure, and (iii) the chemical symbol(s) for the most

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(the answer is not an electron configuration, even though it is) ...

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Choose the best answer and write its letter on the line. 11. The fourth principal energy level has a. 4 orbitals. b. 16 orbitals. c. 32 orbitals. . 9 orbitals. 12. If the electron configuration of an element is $1s^2 2s^2 2p^6 3s^2 3p^5$, the 116 element is a. iron. b. bromine. Core Teaching Resources c. chlorine. d. phosphorus.

CHAPTER 22 Study Guide - Quia

Chemistry is the study of the of matter and the 2. that matter undergoes. ...

Answer the following questions in the space provided. 28. Match each

activity below to one of the five branches of chemistry. ... 14 Core

Teaching Resources Part C Matching

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[05 CTR ch01 7/9/04 3:24 PM Page 7 CHEMISTRY 1](#)

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