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# Definition Of Unsaturated Solution In Chemistry

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## **Solution | Definition of Solution by Merriam-Webster**

An unsaturated solution is a solution that contains less than the maximum amount of solute that is capable of being dissolved. The figure below illustrates the above process and shows the distinction between unsaturated and saturated.

Figure 1. When 30.0 g of NaCl is added to 100 ml of water, it all dissolves, forming an unsaturated solution.

## **Unsaturated | Definition of Unsaturated by Merriam-Webster**

Unsaturated solutions are solutions in which the amount of dissolved solute is less than the saturation point of the solvent (at that specific temperature gradient).

## **Unsaturated Definition in Chemistry - ThoughtCo**

Unsaturated solutions are solutions in which the amount of dissolved solute is less than the

saturation point of the solvent (at that specific temperature gradient). If the amount of dissolved solute is equal to the saturation point of the solvent, the solution is called a saturated solution. To form a mixture, two or more substances must be mixed.

## **What Is an Unsaturated Solution in Chemistry?**

1. Relating to an organic compound in which two or more of the carbon atoms are joined by a double or triple bond and therefore can be combined with additional atoms or radicals. Benzene and unsaturated fatty acids are examples of unsaturated compounds. Compare saturated, monounsaturated, polyunsaturated.

[Unsaturated Solution](#)  
[Wikipedia](#)

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IUPAC definition. According to the IUPAC definition, solubility is the analytical composition of a saturated solution expressed as a proportion of a designated solute in a designated solvent. Solubility may be stated in various units of concentration such as molarity, molality, mole fraction, mole ratio, mass (solute) per volume (solvent) and other units.

**Definition of unsaturated solution - Chemistry Dictionary**

adj. Of or relating to a

solution in which the solvent is capable of dissolving still more of the solute; not saturated. Of or relating to a chemical compound in which all the affinities are not satisfied, so that still other atoms or radicals may be added to it.

*Definition Of Unsaturated Solution In*

Saturated solution definition:

A saturated solution is a solution in which there is so much solute that if there was any... | Meaning, pronunciation, translations and examples

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## **Types of Saturation - Chemistry LibreTexts**

You have an unsaturated solution when there are fewer particles or solutes than solvent in the solution.

Let's break this definition down. Key Terms: Solutes, Solutions, and Polarity A solution is...

[Unsaturated | Definition of Unsaturated at Dictionary.com](#)

unsaturated solution. A solution in which the solvent is able to absorb more solute at a particular temperature. Dictionary of Unfamiliar Words by Diagram Group Copyright © 2008 by Diagram Visual Information Limited. Want to thank

TFD for its existence?

## **Saturated and Unsaturated Solutions | Chemistry for Non-Majors**

Definition of unsaturated solution 1) a solution in which more solute can be dissolved.

### **Unsaturated Solution: Definition & Examples - Video**

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Supersaturated Solution - A supersaturated solution is one in which more solute is dissolved than is necessary to make a saturated solution. A supersaturated solution is unstable solute molecules may crash out of solution given the

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slightest perturbation. Learn more about supersaturated solutions at BYJU'S.

Unsaturated Solutions |

Unsaturated solutions with ...

Saturated Solution. A solution with solute that dissolves until it is unable to dissolve anymore, leaving the undissolved substances at the bottom. Unsaturated Solution. A solution ( with less solute than the saturated solution) that completely dissolves, leaving no remaining substances. Supersaturated Solution.

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Unsaturated Definition and Example Saturated, Unsaturated and Supersaturated Solution | Chemistry

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[Science 101] Saturated and Unsaturated Solution

Unsaturated, Saturated and Supersaturated Solutions

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Angelica Marvie

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Saturated and Unsaturated  
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**Saturated Solutions | Chemistry**

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**Types of Solution - Saturated,  
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**Solution Saturated Solutions**  
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~~solution, unsaturated solution, super saturated solution...~~

~~Explanation in ??????...~~

Saturated and Unsaturated Solution, General Science Lecture | Sabag.pk | Saturated, Unsaturated and supersaturated solution - video clip

Saturated and unsaturated solutions Saturated Solution - Can water dissolve any amount of substance? Class 6 Science An unsaturated solution is a chemical solution in which the solute concentration is lower than its equilibrium solubility. All of the solute dissolves in the solvent. When a solute (often a solid) is

added to a solvent (often a liquid), two processes occur simultaneously. Dissolution is the dissolving of the solute into the solvent.

**Unsaturated - definition of unsaturated by The Free Dictionary**

Definition of unsaturated. : not saturated: such as. a : capable of absorbing or dissolving more of something an unsaturated solution. b : able to form products by chemical addition especially : containing double or triple bonds between carbon atoms unsaturated fats.

Saturated solution definition

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and meaning | Collins ...  
Unsaturated Definition and  
Example Saturated,  
Unsaturated and  
Supersaturated Solution |  
Chemistry

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[Science 101] Saturated and  
Unsaturated Solution  
Unsaturated, Saturated and  
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unsaturated, saturated  
supersaturated **Saturated**  
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~~Solutions—Chapter 6~~

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*Solubility - Wikipedia*

Solution definition is - an action or process of solving a problem. How to use solution in a sentence.

*Unsaturated solution - definition of unsaturated solution ...*

Unsaturated solutions are solutions in which the amount of dissolved solute is less than the saturation point of the solvent (at that specific temperature gradient). If the amount of dissolved solute is equal to the saturation point of the solvent, the solution is called a saturated

solution.

An unsaturated solution is more dilute than a saturated solution. When referring to organic compounds, unsaturated means a molecule contains double or triple carbon-carbon bonds. Examples of unsaturated organic molecules include  $\text{HC}=\text{CH}$  and  $\text{H}_2\text{C}=\text{O}$ . In this context, being saturated can be thought of as being "saturated with hydrogen atoms."