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First of all, we must know that Paper 2 is worth a total of 80 marks and this constitute 50% weightage of your overall Chemistry grade. The time allocated to finish this paper is 1 hour and 45 minutes. According to Singapore Examinations & Assessment Board (SEAB), this paper consists of two sections.

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Paper 2 Organic and Physical Chemistry . 2 *02* IB/M/Jun18/7405/2 Do not write outside the box ... Give your answer to the appropriate number of significant figures. [3 marks]
Concentration of A mol dm⁻³. 0.44 Justify the statement that adding more water to the equilibrium mixture in

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Number of carbon atoms in X = (17 - 4 - 3)/2 = 5 Number of hydrogen atoms in X = (36 - 8 - 6 - 2)/2 = 10 C₅H₁₀ -> Alkene (C_nH_{2n}) 20 D Ethanol burn completely to produce carbon dioxide and water, not carbon monoxide. General formula: C_nH_{2n+1}OH Ethanol can be oxidized by potassium manganate (VII) to form ethanoic acid.

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PAPER 2 1. The flow chart below shows some of the processes involved in large scale production of sulphuric (V I) acid. Use it to answer the questions that follow. a) Describe how oxygen is obtained from air on a large scale (3 marks) (b) (i) Name substance A. (ii) Write an equation for the process that takes place in the absorption chamber. (1mark)

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GUIDELINES ON ANSWERING QUESTIONS IN CHEMISTRY PAPER 2 GUIDELINES ON ANSWERING STRUCTURED QUESTION1. General (a) As with other subjects, you should read all questions carefully and circle all important key words. (b) Answers can be short but they must be precise and complete.

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- There are 100 marks available on this paper.
- The marks for questions are shown in brackets.
- You are expected to use a calculator where appropriate.
- You are reminded of the need for good English and clear presentation in your answers.
- When answering questions 11.3 and 12.2 you need to make sure that your answer:

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