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Chapter 2 Problem 2.19 1/1 Determine the solubility product K_{sp} for $PbSO_4$

. The desired equilibrium is $PbSO_4 \rightleftharpoons Pb^{2+} + SO_4^{2-}$ We can write this a

the sum of two electrochemical equations $PbSO_4 + 2e^- \rightleftharpoons Pb + SO_4^{2-}$ $4 Pb$

$Pb^{2+} + 2e^- \rightleftharpoons Pb$ =

Electrochemical Engineering Principles Solution

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