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Fuller and John N. Harb = 10?14 Chapter 2  
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product  $K_{sp}$  for  $PbSO_4$ . The desired  
equilibrium is  $PbSO_4 \rightleftharpoons Pb^{2+} + SO_4^{2-}$  We  
can write this a the sum of two electrochemical  
equations  $PbSO_4 + 2e^- \rightleftharpoons Pb + SO_4^{2-}$   $Pb^{2+} + 2e^- \rightleftharpoons Pb$  ?

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Examines electrodeposition, redox-flow batteries,  
electrolysis, regenerative fuel cells,  
semiconductors, and other applications of  
electrochemical engineering principles;  
Overlapping chemical engineering, chemistry,  
material science, mechanical engineering, and  
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Electroplating is a general name for processes  
that create a metal coating on a solid substrate  
through the reduction of cations of that metal  
by means of a direct electric current. The part  
to be coated acts as the cathode (negative  
electrode) of an electrolytic cell; the electrolyte  
is a solution of a salt of the metal to be coated;  
and the anode (positive electrode) is usually  
either a block ...

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