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[Electroplating - Wikipedia](#)

Electroplating is a general name for processes that create a metal coating on a solid substrate through the reduction of cations of that metal by means of a direct electric current. The part to be coated acts as the cathode (negative electrode) of an electrolytic cell; the electrolyte is a solution of a salt of the metal to be coated; and the anode (positive electrode) is usually either a block ...

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Examines electrodeposition, redox-flow batteries, electrolysis, regenerative fuel cells, semiconductors, and other applications of electrochemical engineering principles; Overlapping

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Problem 2.19 1/1 Determine the solubility product K_{sp} for $PbSO_4$. The desired equilibrium is $PbSO_4 \rightleftharpoons Pb^{2+} + SO_4^{2-}$. We can write this as the sum of two electrochemical equations $PbSO_4 + 2e^- \rightleftharpoons Pb + SO_4^{2-}$ $Pb \rightleftharpoons Pb^{2+} + 2e^-$ $=?$ $???? \rightleftharpoons$ $=?$

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