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Electroplating is a general name for processes that create a metal coating on a solid substrate through the reduction of cations of that metal by means of a direct electric current. The part to be coated acts as the cathode (negative electrode) of an electrolytic cell; the electrolyte is a solution of a salt of the metal to be coated; and the anode (positive electrode)

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Fuller and John N. Harb =
10?14 Chapter 2 Problem
2.19 1/1 Determine the
solubility product K_{sp} for
 $PbSO_4$. The desired
equilibrium is $PbSO_4 \rightleftharpoons$
 $Pb^{2+} + SO_4^{2-}$ We can
write this as the sum of
two electrochemical
equations $PbSO_4 + 2e^- \rightleftharpoons$
 $Pb + SO_4^{2-}$ $Pb \rightleftharpoons$
 $Pb^{2+} + 2e^-$ $?? = ?$ $???? ? = ?$

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Solution

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reaction fundamentals, and
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and fuel cell characteristics,
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