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becomes the negative electrode,

the cathode. The positive terminal

is attached to the other rod. Free

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caused by electricity in solutions.

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Electrolysis is the process by which ionic

substances are decomposed (broken

down) into simpler substances when an

electric current is passed through them.

Electricity is the flow of electrons or...

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Electrolysis is a chemical reaction caused by

electricity in solutions. Electrolysis can

separate molecules (like separating water into

hydrogen and oxygen gas, which is called

electrolysis of water), electroplate a metal,

can be used for welding, and can even be

used for hair removal. Electrolysis was

discovered by Alessandro Volta.

GCSE Chemistry; Additional Science ISA Chemistry 2.2 ...

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Add Lead (II) Bromide into a beaker and heat so it will turn molten, allowing ions to be free to move to conduct an electric charge. Add two graphite rods as the electrodes and connect this to a power pack or battery. Turn on power pack or battery and allow electrolysis to take place. Negative bromide ions move to the positive electrode (anode) and lose two electrons to form bromine molecules.

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