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Milo Koretsky Department of
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koretsm@engr.orst.edu 2 1.2 An
approximate solution can be
found if we combine Equations
1.4 and 1.5: $k = \frac{V}{m} = \frac{2}{2} \frac{1}{m} = \frac{2}{3} \frac{1}{m}$
 $kT = \frac{2}{3} \frac{1}{m} kT = \frac{2}{3} \frac{1}{m} kT$
Assume the temperature
is

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Chemical and Materials Engineering Cal Poly Pomona (Winter 2009) Contents ...

thermodynamics, we can predict the amount of energy needed to change a system from an equilibrium state to another. For example it will take about 75 kJ to change 1 kg of air at

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