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# Gas Variables Answers

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supplied the highest quality answer. Gas Variables Answers CH301 Worksheet 8—Gases (Answer Key) 1. What do we assume about ideal gases? What is the ideal gas law? Give the units for each variable. Ideal gases are infinitely small, hard spheres that do not interact with each other. They are essentially "blind" to other gas molecules and will bounce off of each other just as they would bounce off a wall.

### **What Is a variable gas - Answers**

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Title Gas Properties Modular Homework Activity: Description  
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The ideal gas law, like any other algebraic equation, allows you to solve for any one variable provided you have the others. R isn't a variable (it's called the ideal gas constant for a reason ...  
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Gas Variables 7 Extension Questions 22. Draw a sample of gas that is colder than all three of the samples in Question 21. Explain why you are sure that it is colder. 23.

Four of the relationships you investigated in Models 1 and 2 are named after scientists who discovered the relationships.

*ChemTeam: Four Gas Law Variables*

Gas pressure is created by the molecules of gas hitting the walls of the container.

This concept is very important in helping you to understand gas behavior.

Keep it solidly in mind. This idea of gas molecules hitting the wall will be used often.

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When the generalized variable of pressure is discussed, the symbol  $P$  is used.

**CHEMISTRY POGIL**  
**ANSWER KEY GAS**  
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In your answer, consider the external and internal pressure data given in Model 1. Read This! Pressure is caused by molecules hitting the sides of a container or other objects. The pressure changes when the molecules change how often or how hard they hit. A nonflexible container is needed if the gas ...  
Gas Variables 7 Extension  
Questions