

General Chemistry Study Guide

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This is a very good companion study guide for the general chemistry book. It arrived quickly and in very good condition. It also has an enhanced media online component that you can access if you purchase the key or if your college/school has a subscription.

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(commonly called the General Chemistry Study Guide) This guide includes 201 pages of information separated into first-term and second-term general chemistry material. Each section contains 8 chapters of material that also aligns to most general chemistry textbooks for a seamless addition to study materials for students.

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Chemistry Overview. The Chemistry examination covers material that is usually taught in a one-year college course in general chemistry. Understanding of the structure and states of matter, reaction types, equations and stoichiometry, equilibrium, kinetics, thermodynamics, and descriptive and experimental chemistry is required, as is the ability to interpret and apply this material to new and ...

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General Chemistry Study Guide
General Chemistry Semester 1 Study Guide 2010 4 A 3p orbital would be higher inenergy and further from the nucleus 7. How does the ground state of an electron differ from the excited state? The ground state is closer to the

nucleus and lower in energy 8. What are the 4 quantum numbers? Principle (1-7), orbital (s,p,d,f), magnetic (p x), spin (?)
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