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to produce a salt, water and heat. 3. A pellet of sodium is sliced in two. 4. Water is heated and changed to steam. £ 5. Potassium chlorate decomposes to potassium chloride and oxygen gas. 6. Iron rusts. 7. Ice melts. 8. Acid on limestone produces carbon dioxide gas. 9. Milk sours. fo r\ . Wood rots. Physical Science IF8767 39 © Instructional Fair, Inc.

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Heat Calculations Heat is measured in units of joules or calories. The amount of heat given off or absorbed can be calculated by the following formula. Solve the following problems. 1. How many calories are absorbed by a pot of water with a mass of 855 g in order to raise the Physical Science Equations / Formulas Flashcards | Quizlet Heat Calculations Physical Science If8767 Heat Calculations Physical Science If8767 Answers Physical Science Equations / Formulas. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. ... the transfer of energy from one physical system to another expressed as the product of a force and the distance through which it moves a body in the direction of that force; "work equals force times distance". ... The amount of heat that must be ...

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heat Potential Energy Raise 1 gram of a substance 1OC. The aluminum has a lower specific heat than the food (specifically the water in the food) and it h erfo atsup n dc lmqiky.A b w will change its temperature even one degree. The temperature will NOT increase during phase changes.

During a phase change, the heat is

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1) Solve for the heat required to increase the water temperature from 33.0 oC to 100.0 oC. Stop here because the water will change phase at this temperature. 2) Solve for the heat required to change the water into steam (no change in temp). 3) Calculate the heat required to change the temperature of the steam from 100.0 oC to 110.0 oC.

Heat Calculations Physical ScienceNew

Solution: Use the following relationships: 2, 3, 4. 1000 mg = 1 g kg Start with the original number and unit. Multiply by a unit factor with the unit to be discarded on the bottom and the desired unit on top. Cancel units. Perform numerical calculations.

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