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Chapter 17 Green chemistry E1. Use the phase diagram for carbon dioxide (page 305) to deduce whether carbon dioxide would be in the solid, liquid or gaseous phase under the following conditions of temperature and pressure: a –100 C and 1 atm pressure b –50 C and 1 atm pressure c –70 C and 10 atm pressure d 0 C and 10 atm pressure AE1.

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10.1 Key questions
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I ' m going into year 11 this year, doing chem 1+2, English 1+2, math methods and specialist 1+2 studio arts 1+2 alongside doing biology 3+4. I really enjoy biology, and it ' s been a really amazing subject, but last year when I was doing 1+2 I found it really difficult to balance my year ten subjects alongside biology.
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Chemistry 2- Chapter 10. STUDY. PLAY. 4 things that we already know about gases. 1. gas is one of the states of matter 2. Temperature and volume of a gas effect its entropy 3. Gases form homogenous mixtures 4. gases contain fast moving particles. entropy.

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Heinemann Chemistry 1 5e CHAPTER 10 REVIEW 10.1 TYPES OF POLYMERS 1 a b 6 c Dipole – dipole bonds 2 aMonomers are small molecules that are able to react to form long chains of repeating units, called polymers. Often contains a carbon – carbon double bond.

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