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Problem Type 3: Given is a mass and unknown is an amount in moles. Mass of given substance (g) Reaction Stoichiometry Problems, continued Section 1 Introduction to Stoichiometry Amount of unknown substance (mol) Amount of given substance (mol) Mass of unknown substance (g) Amount of unknown substance (mol) Amount of given substance (mol) Honors Chemistry Extra Stoichiometry Problems SHORT ANSWER Answer the following questions in the space provided. 1. b The coefficients in a chemical equation represent the (a)

masses in grams of all reactants and products.

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Chapter 9

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Extra Stoichiometry Problems 1. Silver nitrate reacts with barium chloride to form silver chloride and barium nitrate. a. Write and balance the chemical equation. $2 \text{AgNO}_3 + \text{BaCl}_2 \rightarrow 2 \text{AgCl} + \text{Ba}(\text{NO}_3)_2$ b. If 39.02 grams of barium chloride are reacted in an excess of silver nitrate, how many ... Extra Practice - Stoichiometry Answers Author ... [chapter 9 review stoichiometry section 3 answers modern](#) ... The Stoichiometry chapter of this Holt McDougal Modern Chemistry Companion Course helps students learn the essential chemistry lessons of stoichiometry.

Section 1

Introduction to Chapter 9

Stoichiometry

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... A balanced chemical equation may be used to describe a

reaction's stoichiometry (the relationships between amounts of reactants and products). Coefficients from the equation are used to derive ... 7.3:

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Concept Review: Calculating Quantities in Reactions 1. mole 2. balance 3. relative 4. coefficients 5. molar mass 6. liquids 7. density 8. Avogadro's number [mc06sete cFMsg i-vi - Ed W. Clark High School](#)

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Section Quiz: Gas Volumes and the Ideal Gas Law In the space provided, write the letter of the term or phrase that best completes each sentence or best

answers each question. spirit. Although the
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