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Answers to Gram
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bottom is says:
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Gram Formula

Mass Instructional Fair Answer Key NAMING HYDR O CARBONS Name Name the compounds below according to the IUPAC naming system 1. H H H I I H —C —C —C —H I I I H H H 5.

IDEAL GAS LAW Use the ideal Gas Law below to solve the following problems. pressure in atmospheres volume in liters number of moles L atm Universal Gas Constant = 0.0821

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percentage
composition: vapor.

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Naming Ionic
Compounds - Ms.
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enstructional Fair,
Inc. 5. 35 g of
 $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$
Determine the
number of grams in
each of the
quantities below. 1,
2.5 moles of NaCl 2.
0.50 moles of 3, 1.70
moles of KMnO_4 4,
0.25 moles of KCl e,
ST 5. 3.2 moles of
 $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$
Chemistry IF87 ó ó
appohigh.org
Naming Ionic
Compounds
Directions: Name
the following
compounds using
the Stock Naming
System. 1. CaCO_3
... Chemistry

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products provide
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building exercises
designed to help
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and inspire
students from
kindergarten to 8th
grade. Working in
collaboration with
educators,
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reaches beyon.
[www.newburypark
highschool.net](http://www.newburyparkhighschool.net)

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 Composition states that the
 Worksheet Answers volume of a gas
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Homework- assuming that
Chemistry: pressure is constant.
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9781568221434 ... formulas: - or $x = x$
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 first in their family to FIGURES Name
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 average undergrad at Solutions 1) If I have
 MSU Denver is 25 5.6 liters of gas in a
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 transfer to MSU compress the gas
 Denver. Take your until its volume is 4.8
 first steps on campus

L, what will the new
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 piston be?
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 Writing Formulas
 From Names
 Directions: Write
 the formulas of the
 following
 compounds. 1.
 ammonium
 phosphate (NH 4)
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 Law 22 Solubility
 (Polar vs.

Nonpolar) 74
Periodic Table
Worksheet 36
Acids and Bases
Crossword 90.
Periodic Gram
Formula Mass 49
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CALCULATIONS

USING

SIGNIFICANT

FIGURES Name ?

When multiplying and

dividing, limit and

round to the least

number of significant

figures in any of the
factors. 11 Example 1 :
 $23.0 \text{ cm} \times 432 \text{ cm} \times 19$
 $\text{cm} = 188,784 \text{ cm}^3$ 11