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89. Geometric isomerism It is possible for single C-C bonds to rotate freely, however, double C=C bonds cannot. Thus if the two carbon atoms of a C=C bond carry different groups, it becomes possible to orientate these groups in two ways to create geometric isomers.

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He co-authored seven editions of INTRODUCTION TO GENERAL, ORGANIC, AND BIOCHEMISTRY, ORGANIC CHEMISTRY, and several Laboratory Manuals. He is also the author of EXPERIMENTAL PHYSICAL CHEMISTRY and co-author of numerous monographs and research articles. Professor Bettelheim received his Ph.D. from the University of California, Davis.

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About the Author Felix Lee and William H. Brown are the authors of Student Solutions Manual to accompany Introduction to Organic Chemistry, 6e, published by Wiley.

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