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The Laboratory Manual is a reference manual for ORA laboratory personnel and supporting units.

Laboratory Manual for Principles of General Chemistry ...

The equation for the hydrolysis reaction is,trans-[Co(en)2Cl2]+ + H₂O

cis/trans-[Co(en)2(H2O)CI]+ +Cl(dark green) (burgundy to red violet) The reactant is green while the product is supposed to be burgundy red to violet, but because ofproblems we have had in recent years, the product is sometimes yellow or orange.

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CHEMISTRY 121 LABORATORY (Fall

teaching and learning experience available 2013) Beforecoming to laboratory you should read the Exercise scheduled for the week, including the "Assigned Reading, " and complete the Homework section of the Exercise. Each week there will be a quiz. The quiz will cover both the laboratory exercise and homework from the previous week as well as the homework for the current week. **UBCO COURSE PACK CHEM** 111/113 121/123

> General Chemistry I Laboratory Manual Winter term 2011-12 Lab begins the first week of classes Required Text for CHEM 122 (All sections) You must bring this lab manual, plus safety glasses, to the first lab period. Safety glasses are required, and they are sold at the bookstore.

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CHEM 121- Exp #34- Lab DataChem 121 Density Lab Chem 121 Iron Sulfate Synthesis Lab Chem 121 Lab 8 Chem 121 Lab Worksheet #1

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> In a 100 mL beaker, add 5 mL of a 5% aqueous solution of hexamethylenediamine (I,6- diaminohexane) and 5 drops of 20% sodium hydroxide solution. Carefully add 5 mL of a 5% solution of sebacovi chloride (decanediovi chloride) in cyclohexane to the solution by pouring it down the wall of the slightly tilted beaker.

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