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SOLUTIONS CHAPTER-6  
PART-1 6.1 Before  
solving this  
question, examine  
ex.6.1 in Lamarsh and  
for cross sections  
**Introduction to  
Nuclear Engineering  
(4th Edition): John R  
...**  
A glass of water is  
known to contain  $6.6 \times 10^{24}$  atoms of  
hydrogen. How many  
atoms of deuterium  
( $2H$ ) are present?  
Solution. According  
to Table 11.2 in

Appendix II, the  
isotopic abundance of  
 $2H$  is 0.015 a/o. The  
fraction of the  
hydrogen, which is  
 $2H$ , is therefore  $1.5 \times 10^{-4}$  The  
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Section 6.2.  
Solutions for  
Section 6.3. ...  
Convert P to a CFG,  
and then convert  
the CFG to a PDA,  
using the two  
constructions given  
in Section 6.3. The  
result is a one-  
state PDA

equivalent to P.  
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 CHAPTER-6 PART-1. 6.1.  
 Before solving this  
 question,examine ex.6.1  
 in Lamarsh and for  
 cross sections and  
 other constants use  
 table 6.1; Pu 0.03 Na  
 32.33 and then; Pu Na  
 Pu Pu 0.03 Na 32.33 Pu  
 Na Pu 1 1 f 0.108 and  
 2.61 N Na a , Na Na \*  
 Navgdro 1 a , Na N Pu a  
 , Pu M Na 1 Pu \*  
 Navgdro a , Pu M Pu a ,  
 Pu 2.11b a , Na 8e 4b  
 and finally k f  
 2.222(sup ercritical )  
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perspectives and  
approaches to  
change management,  
but also greater  
confidence in my  
abilities to ensure  
change readiness  
for the ...