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Express your answer using a chemical  
symbol. Hg. Part A Arrange the  
following oxides in order of increasing  
acidity. Rank from least acidic to most  
acidic. To rank items as equivalent,  
overlap them. CaO, Al<sub>2</sub>O<sub>3</sub>, SiO<sub>2</sub>, CO<sub>2</sub>,  
P<sub>2</sub>O<sub>5</sub>, SO<sub>3</sub>. Part A barium oxide with  
water Express your answer as a  
chemical equation. Identify all of the  
phases in your answer.

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following elements: Kr, Ga, Br, As, K. -  
Answer: Kr (8), Br (7), As (5), Ga (3), K (1).  
Valence electrons are equal to the atom's  
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