

under given conditions. (of an organic compound) containing no double or triple bonds; having each single bond attached to an atom or group. (of an inorganic compound) having no free valence electrons.

[Saturated solution definition and meaning | Collins ...](#)

1. Relating to an organic compound in which two or more of the carbon atoms are joined by a double or triple bond and therefore can be combined with additional atoms or radicals. Benzene and unsaturated fatty acids are examples of unsaturated compounds. Compare saturated, monounsaturated, polyunsaturated.

[SATURATED SOLUTION | meaning in the Cambridge English ...](#)

Supersaturated Solution - A supersaturated solution is one in which more solute is dissolved than is necessary to make a saturated solution. A supersaturated solution is unstable solute molecules may crash out of solution given the slightest perturbation. Learn more about supersaturated solutions at BYJU'S.

Types of Saturation - Chemistry LibreTexts

[Saturated Solution: Definition & Examples - Video & Lesson ...](#)

The uranium solution was a twofold dilution of a saturated solution. From the Cambridge English Corpus The stain was prepared as a saturated solution in pure isopropanol.

Definition of unsaturated solution - Chemistry Dictionary

Unsaturated solution Definition from Encyclopedia Dictionaries & Glossaries. Wikipedia Dictionaries. English Wikipedia - The Free Encyclopedia. In chemistry, saturation (from the Latin word saturare, means to fill) has diverse meanings, all based on reaching a maximum capacity.

Unsaturated solution - definition of unsaturated solution ...

adjective. not saturated; having the power to dissolve still more of a substance. Chemistry. (of an organic compound) having a double or triple bond and capable of taking on elements or groups by direct chemical combination without the liberation of other elements or compounds, as ethylene, $\text{CH}_2=\text{CH}_2$; undersaturated.

Saturated | Definition of Saturated at Dictionary.com

An unsaturated solution is a chemical solution in which the solute concentration is lower than its equilibrium solubility. All of the solute dissolves in the solvent. When a solute (often a solid) is added to a solvent (often a liquid), two processes occur simultaneously.

[Unsaturated Solutions | Unsaturated solutions with ...](#)

The definition of solute concentration is the amount of solutes or particles that are dissolved in a solution. So would an unsaturated solution be able to have more solutes dissolved in the

solution?

Meaning Of Unsaturated Solution

A solution that is unsaturated does not have excess material or solvent within the liquid. Unsaturated solutions have the potential to effectively dissolve more material before reaching the point of full saturation. A saturated solution is as saturated as it can possibly be under normal conditions.

[Unsaturated - definition of unsaturated by The Free Dictionary](#)

Unsaturated fat or oil is either monounsaturated or polyunsaturated, found in plants, vegetable oil, and fish, and thought to be better for your health than saturated fat. containing less of a substance than can be absorbed, and therefore not in a state of balance

Saturated solution definition: A saturated solution is a solution in which there is so much solute that if there was any... |

Meaning, pronunciation, translations and examples