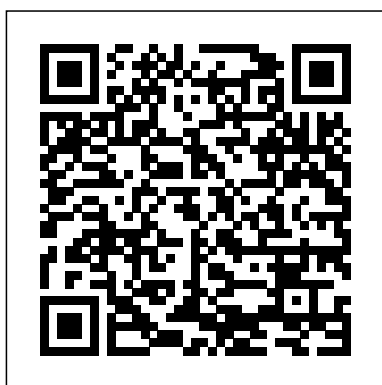

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Aqueous Solutions and Colligative Properties
Section 1- Compounds in Aqueous Solutions
dissociation is the separation of ions that
occurs when an ionic compound dissolves see
sample problem A on page 436 do practice
problem #1 on page 436 Solubility
precipitation reactions occur when
combinations of ions in a solution have an
extremely low solubility and a ...

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An easier way to understand reactions is to
once again consider cooking, but this time,
think about making a pizza Modern
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The foundation of pizza is, of course, the
crust, and pizza dough is the raw . . .

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CHAPTER 13 REVIEW Ions in Aqueous Solutions and Colligative Properties

MIXED REVIEW SHORT ANSWER

Answer the following questions in the space provided. 1. Match the four compounds on the right to their descriptions on the left.

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CHAPTER 12 REVIEW Solutions

SECTION 3 SHORT ANSWER Answer

the following questions in the space provided. 1. Describe the errors made by the following students in making molar solutions. a. James needs a 0.600 M solution of KCl. He measures out 0.600 g of KCl and adds 1 L of water to the solid.

CHAPTER 12 REVIEW Solutions

CH 1 Reading Assignment Modern

Chemistry. CH 1 Vocabulary-New. CH 1

Mixed Questions. CH 1 Matter & Energy Vocabulary-New ... CH 10 Section Review

13.1 & 13.2 . CH 10 Section Review 13.3

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13 . Reference: Review Activities:

Colligative Properties Logic Puzzle.

13 Ions in Aqueous Solutions and Colligative Properties

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Assessment Chapter Test B

CHAPTER 6 REVIEW Chemical Bonding

MIXED REVIEW SHORT ANSWER

Answer the following questions in the space provided. 1. a.

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Modern Chemistry 92 Chapter Test Name

Class Date Chapter Test B, continued 31.

Calculate the energy released by freezing

13.8 g of a liquid. Its molar mass is 82.9

g/mol, and its molar enthalpy of fusion is

4.60 kJ/mol. PART VI Define the term

equilibrium vapor pressure. Use it to explain the follow-ing two phenomena.

CHAPTER 13 REVIEW Ions in

Aqueous Solutions and ...

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6 Chemical Bonding

The Ions in Aqueous Solutions and

Colligative Properties chapter of this Holt

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Modern Chemistry 113 Ions In
Aqueous Solutions And Colligative
Properties CHAPTER 13 REVIEW

Ions in Aqueous Solutions and
Colligative Properties SECTION 2
PROBLEMS Write the answer on
the line to the left. Show all your
work in the space provided. 1. _____

a. Predict the boiling point of a
0.200 m solution of glucose in
water.

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13. Colloid. Electrolyte.