

## Modern Chemistry Chapter 21 Section 1 Review Answers

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### CHAPTER 21 REVIEW Nuclear Chemistry

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Modern Chemistry 175 Nuclear Chemistry CHAPTER 21 REVIEW Nuclear Chemistry SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. Match each of the following statements with the process(es) to which they apply, using one of the choices below: (1) fission only (3) both fission and fusion

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Modern Chemistry 179 Organic Chemistry CHAPTER 22 REVIEW Organic Chemistry SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. Name two types of carbon-containing molecules that are not organic. \_\_\_\_ 2. \_\_\_\_ Carbon atoms form bonds readily with atoms of

### 5 The Periodic Law

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Chemistry Predict the mode of decay of (a) carbon-14, (b) (b) xenon-118. (b) Xenon has an atomic number of 54. Thus, xenon-118 has 54 protons and  $118 - 54 = 64$  neutrons, giving it a neutron-to-proton ratio of According to Figure 21.2, stable nuclei in this region of the belt

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CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. The element boron, B, has an atomic mass of 10.81 amu according to the periodic table.

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CHAPTER 1 REVIEW Matter and Change MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Classify each of the following as a homogeneous or heterogeneous substance. a. sugar d. plastic wrap b. iron filings e. cement sidewalk c. granola bar 2. For each type of investigation, select the most appropriate branch of chemistry from the following

Chemistry Chapter Resources. Chapter One Matter & Change.

Objectives: Chapter 1 Honors: ... CH 1 Reading Assignment Modern Chemistry. CH 1 Vocabulary-New. CH 1 Mixed Questions. ... CH 21 Section Review 3 & 4. CH 21 Test Wksh. CH 1 White Board Practice. CH 3 Nuclear Processes.

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CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain similarities in the ...

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