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CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter SECTION 2 SHORT

ANSWER Answer the following questions in the space provided. 1. In cathode-ray tubes, the cathode ray is emitted from the negative electrode, which is called the cathode . 2. The smallest unit of an element that can exist either alone or in molecules containing the

Modern Chemistry Chapter 2-3 Review Answers

CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER

Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to

Mendeleev ' s original design. ... 40 THE PERIODIC LAW MODERN CHEMISTRY ...

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CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter SECTION 2 SHORT

ANSWER Answer the following questions in the space provided. 1. In cathode-ray tubes, the cathode ray is emitted from the negative electrode, which is called the cathode . 2. The smallest unit of an element that can exist either alone or in molecules containing the

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MODERN CHEMISTRY The atomic number of nickel-60 is 28. How many neutrons does this iso- tope have?

Carbon-14 has 8 neutrons. What is the

atomic number of carbon-14? 11 CHAPTER 3 TEST HR W material copyrighted under notice appealing earlier in this work.

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CH 1 Reading Assignment

Modern Chemistry. CH 1

Vocabulary-New. CH 1 Mixed Questions. CH 1 Matter &

Energy Vocabulary-New ...

Chapter 3 CH 3 Atomic

Structure. Reference: CH 3

Atomic Structure Organizer.

... Ch 6 Section Review 6.3.

Ch 6 Section Review 6.4. Ch 6

Section Review 6.5.

7 Chemical Formulas and Chemical Compounds

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CHAPTER 3: ATOMS: THE BUILDING
BLOCKS OF MATTER Include
graphic organizer(s) for this
chapter The following pages
contain the bulk (but not all)
of the information for the
chapter 3 test. Focus on this
content, but make sure to
review class notes,
activities, handouts,
questions, etc. If you study
this document and NOTHING
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3:3 mol H₂; or their reciprocals 7. Given the following equation: 4NH₃(g) + 6NO(g) → 5N₂(g) + 6H₂O(g) 1 mol NO:1 mol H₂O a. What is

the mole ratio of NO to H₂O? 3 mol NO:2 mol NH₃ b. What is the mole ratio of NO to NH₃? 0.360 mol c. If 0.240 mol of NH₃ react according to the above equation, how many moles of NO will be consumed? 8.

5 The Periodic Law

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shipping on qualifying offers. Modern Chemistry : Chapter Tests with Answer Key Modern Chemistry: Chapter 3 Review Flashcards | Quizlet CHAPTER 10 REVIEW States of

Matter SECTION 3 SHORT ANSWER
Answer the following questions
in the space provided. 1. Match
description on the right to the
correct crystal type on the
left. b ionic crystal (a) has
mobile electrons in the crystal
c covalent molecular crystal (b)
is hard, brittle, and
nonconducting a metallic crystal
(c) typically has the lowest
melting point of the four
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...
CHAPTER 6 REVIEW Chemical

Bonding SECTION 3 SHORT ANSWER
Answer the following questions
in the space provided. 1. a The
notation for sodium chloride,
NaCl, stands for one (a) formula
unit. (c) crystal. (b) molecule.
(d) atom. 2. d In a crystal of
an ionic compound, each cation
is surrounded by a number of (a)
molecules. (c) dipoles. (b ...
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Review
The Atoms: The Building
Blocks of Matter chapter of
this Holt McDougal Modern
Chemistry Companion Course
helps students learn the
essential lessons associated
with atoms.

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Modern Chemistry Chapter 6

Review Answers Section 3

SECTION 1 continued 6. Some binary compounds are ionic, others are covalent. The type of bond favored partially depends on the position of the elements in the periodic table.

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6 Chemical Bonding

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Chemistry and Physics of
Modern Materials

2 Measurements and Calculations

3. Compare and contrast a model with a theory. Theories are broad generalizations used to explain observations. Models are a physical object used to illustrate or explain complex concepts or an explanation of how phenomena occur and how data and events are related.

MODERN CHEMISTRY MEASUREMENTS
AND CALCULATIONS 9

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