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# Modern Chemistry Chapter 5

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a. 2 and 4 c. 4 and 5 b. 3 and 4 d. 5 and 6 \_\_\_\_ 21. The discovery of the noble gases changed Mendeleev's periodic table by adding a new. a. period. c. group. b. series. d. sublevel block. \_\_\_\_ 22. In the modern periodic table, elements are

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ordered according to. a. decreasing atomic mass. c. increasing atomic number. b. Mendeleev's original ...

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Modern Chemistry 37 The Periodic Law CHAPTER 5 REVIEW The Periodic Law SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. \_\_\_\_\_

When an electron is added to a neutral atom, energy is (a) always absorbed.

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CHAPTER 5 REVIEW The Periodic Law SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. c When an electron is added to a neutral atom, energy is (a) always absorbed. (c) either absorbed or released. (b) always released. (d) neither absorbed

nor released. 2. d The energy required to remove an electron from a neutral atom is the atom ' s

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a 9. lanthanides 10. 2 11.

fourth 12. transition elements  
13. 32 14. valence electrons  
15. electron affinity 16.  
electronegativity 17. ionization  
energy 18. 3 s<sup>2</sup> 3p<sup>4</sup> 19. atomic  
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