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CHAPTER 6 REVIEW Chemical Bonding SECTION 4
SHORT ANSWER Answer the following questions in the space provided Modern chemistry chapter 6 review worksheet answers. 1. b In metals, the valence electrons are considered to be (a) attached to particular positive ions. (c) immobile. (b) shared by all surrounding atoms. (d) involved in covalent bonds.
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questions related to electrolysis and electroplating, questions include classification of electrolyte, molten and aqueous solutions, flow of electrons, the product selection at the anode and cathode based on position in the electrochemical series, the concentration of the electrolyte and type of electrodes used, constructing chemical equations and ...

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Modern Chemistry Chapter 6 Chemical Bonding Test Answer ...

Answers Modern Chemistry Chapter 6 Worksheet 6 Chemical Bonding CHAPTER 6 REVIEW Chemical Bonding SECTION 3 SHORT ANSWER Answer the following questions in the space provided 1 a The notation for sodium chloride, NaCl, stands for one (a) formula unit (c) crystal (b) molecule (d) atom 2 d In a crystal of an
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dipole. a molecule or a part of a molecule
that contains both positively and negatively
charged regions. hydrogen bonding. the
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hydrogen atom that is bonded to a highly
electronegative atom of one molecule is
attracted to two unshared electrons of
another molecule.

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