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the following questions in the space provided. 1. a.

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left. Show all your work in the space provided. 1. 88% The actual yield of a reaction is 22 g and the

theoretical yield is 25 g. Calculate the percentage yield. 2. 6.0 mol of N₂ are mixed with 12.0 mol of

H₂ according to the following equation: N₂(g) + 3H₂(g) ? 2NH₃(g) N

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